

0040-4020(94)00851-5

Total Synthesis of Natural (+)-Spatol. Confirmation of The Absolute Stereostructure¹

Masahide Tanaka, Kiyoshi Tomioka† and Kenji Koga*

Faculty of Pharmaceutical Sciences, University of Tokyo, Hongo, Bunkyo-ku, Tokyo 113, †Institute of Scientific and Industrial Research, Osaka University, Ibaraki, Osaka 567, Japan

Abstract: Total synthesis of optically pure spatol (+)-1 was achieved by employing asymmetric (2+2) photocycloaddition of the optically pure butenolide 12 with cyclopentenone ethylene ketal. The two major photocycloadducts 14, 15 were separately converted into the cis, anti, cis-tricyclo- $[5.3.0.0^{2,6}]$ decane 9 and then to the sulfonium 11. Coupling of 11 through ylide with optically active epoxyaldehyde 24 provided optically pure 1 with definite absolute configuration.

(+)-Spatol 1 is a representative of spatane diterpenes isolated from brown algae of south pacific ocean.² Spatane diterpenes are characterized by unique tricyclo[$5.3.0.0^{2.6}$]decane ring,³ which is also found in bourbonene sesquiterpenes 3, 4⁴ with the opposite absolute configuration. Furthermore, the tricyclic ring of spatol is attached by a vinyl diepoxide carbon chain with three contiguous chiral centers.⁵

$$H \stackrel{\text{Me}}{\longrightarrow} OR^{2} \quad 1: R^{1} = \stackrel{O}{\longrightarrow} OR^{2} = H \\ 2 R^{1} = \stackrel{O}{\longrightarrow} R^{2} = H \\ 5: R^{1} = \stackrel{O}{\longrightarrow} OH \\ R^{2} = 4 \text{-BrPhCO}$$

The stereochemistry of 1 was elucidated based on the X-ray crystallography of its bromobenzoate derivative 5, in which one of the two epoxide rings was opened by chloride anion during derivatization.² However, the corresponding configuration was tentatively assigned to S assuming S_N^2 opening of the epoxide with chloride anion to 5. Along with these intriguing structural features, the potent cytotoxicity and limited supply from the natural source attracted much concern of synthetic chemists.⁶

We have already succeeded in the total synthesis of optically pure (+)-stoechospermol 2, another representative of spatane diterpenes,⁷ and bourbonene sesquiterpenes 3, 4⁸ utilizing intramolecular and intermolecular asymmetric (2+2) photocycloadditions. In these syntheses, *cis,anti,cis*-tricyclo[5.3.0.0^{2,6}]-decane skeleton was constructed in both configurations starting from (S)- γ -hydroxymethyl- γ -butenolide 6⁹ by the application of the intramolecular or intermolecular mode of the photocycloaddition. Encouraged by these successful approaches, we focused our efforts on the total synthesis of optically pure spatol 1, which is a subject of the present full article.¹⁰

Synthetic Strategy

The synthesis of spatol 1 was essentially based on the same strategy applied for 2.7 Asymmetric (2+2) photocycloaddition of 7 or 12 provides optically pure *cis,anti,cis*-tricyclo[5.3.0.0^{2,6}]decane 9, which is

M. TANAKA et al.

convertible to an allylic alcohol 10 according to the already developed procedure.⁷ Then, the allylic alcohol 10 is converted to the corresponding sulfonium $11.^{11}$ Coupling of 11 *through* sulfonium ylide with an optically active epoxy aldehyde with definite absolute configuration¹² would provide optically pure spatol 1 with definite absolute configuration.



Preparation of the Key Intermediate 9

The intramolecular photocycloaddition approach is highly stereoselective, however, requires somewhat long route to reach 9.7 For the synthesis of 1, we adapted the intermolecular photocycloaddition approach. The starting butenolide (R)-6, prepared from readily available (S)-6 by the inversion of configuration,¹³ was converted to (+)-12 according to the previously developed procedure.¹⁴ The key (2+2) photocyclo-addition of (+)-12 with 13 was conducted in acetonitrile under irradiation with low pressure mercury lamp to provide the four regio- and stereoisomeric photoadducts.⁷ Acidic hydrolysis of these photoadducts followed by careful separation afforded the desired tricyclic ketones, (+)-14 and (-)-15 in 14% and 28% yield, respectively. The minor and undesired two stereoisomers were also isolated in 9% and 11% yields.⁷



a) hv (254nm), CH₃CN; b) 10%H₂SO₄, acetone 14: 28%, 15: 14%

Wittig olefination of 14 followed by stereoselective hydrogenation afforded 17 in 60% yield, which was then converted to 9 in four steps (i. MeLi, ii. LiOMe, iii. NaIO₄, iv. NaOH) in 76% overall yield.⁷



a) CH₃PPh₃Br, n-BuLi, DME, 60%; b) H₂, 10%Pd-C, AcOH, quant; c) MeLi, THF; LiOMe, MeOH-THF; NalO₄, aq. AcOEt; NaOH, aq. MeOH, 76%; d) PhSeCl, AcOEt; H₂O₂ pyridine-CH₂Cb₂, 69%; e) CH₂N₂, THF; f) toluene, heat, 75%; g) H₂, 10% Pd-C, AcOEt, 93%; h) ethanedithiol, BF₃OEt₂, CH₂Cb₂, 92%; i) Raney Ni (W-4), EtOH, 95%

The conversion of 15 into 9 was achieved through 17. Phenylselenylation and oxidation¹⁵ of 15 provided 18, which was then treated with diazomethane and heated in toluene to give 19. Then, stereo-

selective hydrogenation of 19, and deoxygenation via thioacetal provided 17, which was identical with that

The optically pure key intermediate 9 was then converted into 10 by the procedure developed before.⁷

Model Study of the Vinylic Diepoxide Synthesis

obtained from 14.

Since the synthetic methodologies for linear and contiguous vinylic diepoxide systems are quite few,⁵ we investigated model reaction of three sulfonium salts 23 with racemic epoxyaldehyde 24. The sulfonium salts were prepared from 20^{16} as shown.



Treatment of 23a with LDA or BuLi in DME followed by 24, however, resulted in the formation of 28a as a major product by preferential deprotonation at the methylthio group and following [2,3]-sigmatropic rearrangement. Attempted reaction by diphenyl sulfonium salt 23c was also unsatisfactory to give rise to a mixture of diepoxides in low yield as shown in the Table 1. Fortunately, it was found that, when the diethyl sulfonium salt 23b was treated with methyllithium as a base in the presence of 1 eq. of HMPA in DME followed by 24, the desired vinylic diepoxide 25 was obtained as a major product in 25% yield among the possible other three isomers 26 and 27 (a mixture of 5:1 *trans* isomers) and 28b.



entry	23	R	Base	25/%	26/%	27ª/%	28/%
1	23a	Me	LDA	0	0	0	 74b
2			n-BuLi	4	0	5	40 ^b
3	23c	Ph	LDA	3	0	2	-
4			n-BuLi	3	2	4	-
5	23b	Et	n-BuLi	11	5	17	19c
6			MeLi	20	5	19	7¢
7			MeLi + HMPA (1 eq)	25	4	22	11c

Table 1. Coupling of 23 with 24

a) A mixture of 5:1 two trans-diastereomers. b) R'=H. c) R'=Me.

The structure of 25 was assigned based on the following experiments. The syn, syn-alcohol 30 (J_{Ha-Hb} = 8 Hz),¹⁷ prepared by coupling of allyl stannane 29 with 24 according to the established chemistry of Yamamoto and Keck,¹⁸ was treated with Me₃O+-BF₄- in dichloromethane and then with methyllithium in DME-HMPA to afford a *cis*-diepoxide 26 (J_{Ha-Hb} = 4 Hz)¹⁹ having the undesired relative configuration. Coupling constant between the protons of the newly created epoxide of 25 is 5 Hz and this indicates that 25 has the desired relative configuration identical with that of 1.



Total Synthesis of Optically Pure Spatol

The allylic alcohol 10 was converted to the crystalline sulfonium salt 11 in 79% overall yield in four steps (i. MsCl-Et₃N/CH₂Cl₂; ii. NaSEt/DMF; iii. 10% H₂SO₄/acetone; iv. Et₃O·BF₄/CH₂Cl₂). The optically active epoxyaldehyde (R)-24 was prepared from the corresponding known (S)-epoxyalcohol¹² by the Collins oxidation.



a) MsCl, Et₃N, CH₂Cl₂; EtSNa, DMF, 89%; b) 10%H₂SO₄ acetone, 92%; c) Et₃O+BF₄ CH₂Cb, 75%

Treatment of 11 with methyllithium in the presence of 1 eq. of HMPA in DME at -70 °C for 1h and then with optically active (R)-24 at -70 °C for 1h and then at rt for 1 h afforded, after careful separation by HPLC (Waters Radial Pak B, acetone-hexane/1:6), (+)-spatol 1 and two isomers 32, 33^{20} in a ratio of 24:12:64 in 13% yield and the rearranged product 34 in 35% yield.



Optical rotation, melting point, tlc behavior, and spectroscopic data of the synthetic spatol 1 ($[\alpha]_D^{a_3}$ +45.5 °(CHCl₃), mp 100-101 °C, mmp 100-102°C) were completely identical with those of the natural spatol ($[\alpha]_D$ +45.6 °(CHCl₃), mp 100-102 °C), kindly provided by Dr. Fenical.² Since the optically pure epoxy aldehyde 24 with the definite absolute configuration was incorporated in 1, the stereochemistry of natural spatol was firmly established.

Conclusion

The first total synthesis of (+)-spatol 1 with natural configuration was achieved by utilizing the intermolecular (2+2) photocycloaddition reaction for the construction of tricyclic skeleton and the reaction of sulfonium ylide with optically active epoxy aldehyde for the vinylic diepoxide moiety. The synthesis also served the direct evidence for the terminal epoxide stereochemistry to be S-configuration in natural spatol.

Experimental²¹

(+)-3-Methyl-4-pivaloyloxymethyl-2-methyl-2-buten-4-olide 12 Based on the reported procedure, 13,14 12 was prepared as colorless needles of mp 64-66 °C (AcOEt-hexane). $[\alpha]_D^{20}$ +68.9°(c 1.05, CHCl₃).

Photocycloaddition of (+)-12 with 13 According to the procedure described before, a mixture of (+) 12 (3.95 g, 18 mmol) and 13 (18 mL, 0.16 mol) was irradiated by low pressure mercury lamp, and the products were separated. The first fraction contained diastereomeric ketal of 14 (1.6 g) as colorless oil. Purification was achieved by converting into the ketone. The second fraction contained diastereomeric ketal of 15 (0.62 g, 10%) as colorless needles of mp 114.5-115 °C (AcOEt-hexane). $[\alpha]_D^{20}$ -92.7 °(c 0.43, CHCl₃). Spectroscopic data were identical with those of its enantiomer.⁷ The third fraction contained ketals of 15 and 14 as a colorless oil (4.3 g). Crystallization from AcOEt-hexane gave ketal of 14 as colorless needles (1.40 g, 22%) of mp 126-128 °C. $[\alpha]_D^{20} + 22.9$ °(c 0.35, CHCl₃). Spectroscopic data were identical with those of 15 and 14, contained in mother liquor (1.85 g), was achieved by converting into ketones as described below.

(+)-(15,2R,5R,6S,7R)-6-Methyl-5-pivaloyloxymethyl-4-oxatricyclo[5.3.0.0^{2,6}]decan-3,8-dione A solution of diastereomeric ketal of 14 (1.60 g, see above) and 10% H₂SO₄ (1.5 mL) in acetone (15 mL) was heated under reflux for 1 h. After neutralization with satd. NaHCO₃, the mixture was extracted with CH₂Cl₂ (20 mL x 3), and the combined extracts were dried. Concentration and column chromatography (AcOEt-hexane 1:2) afforded colorless needles of mp 83.5-84.5 °C (AcOEt-hexane) (0.61 g, 11% form (+)-12). $[\alpha]_D^{20}$ +37.4 °(c 0.93, CHCl₃). IR (KBr): 1775, 1720 cm⁻¹. NMR δ : 1.24 (9H, s, (CH₃)₃), 1.28 (3H, s, CH₃), 1.9-2.6 (4H, m, (CH₂)₂), 2.7-3.1 (3H, m, CH), 4.2-4.5 (3H, m, OCH₂CHO). MS m/z: 295 (M⁺+1). Anal. Calcd for C₁₆H₂₂O₅: C, 65.29; H, 7.53. Found: C, 65.03; H, 7.39.

(+)-(1*R*,2*S*,5*R*,6*R*,7*S*)-6-Methyl-5-pivaloyloxymethyl-4-oxatricyclo[5.3.0.0^{2,6}]decan-3,10dione 14 and (-)-(1*R*,2*S*,5*R*,6*R*,7*S*)-6-methyl-5-pivaloyloxymethyl-4-oxatricyclo[5.3.0.0^{2,6}]decan-3,8-dione 15 A mixture containing ketals of 15 and 14 (1.85 g, obtained from the above mother liquor) and 10% H₂SO₄ (10 mL) in acetone (50 mL) was heated under reflux for 1h. After neutralization with satd. NaHCO₃, the mixture was concentrated, taken up into CH₂Cl₂ (50 mL), washed with water, and dried. Concentration and column chromatography (AcOEt-hexane 1:2) afforded (+)-14 (0.35 g, 6% form (+)-12) as colorless needles of mp 132-133 °C (AcOEt-hexane) and (-)-15 (0.76 g, 14% from (+)-12) as colorless plates of mp 110-112 °C (AcOEt-hexane). (+)-14: $[\alpha]_{20}^{20}$ +154 °(c 0.96, CHCl₃). IR (KBr): 1775, 1740 cm⁻¹. NMR δ : 1.17 (9H, s, (CH₃)₃), 1.20 (3H, s, CH₃), 1.9-2.3 (2H, m), 2.3-2.7 (2H, m), 2.7-2.8, 2.8-2.9, 3.0-3.2 (each 1H, m), 4.04, 4.36 (each 1H, dd, J=2, 12 Hz, OCH₂CHO), 4.55 (1H, t, J=2 Hz, CH). MS m/z: 295 (M⁺+1). Anal. Calcd for C₁₆H₂₂O₅: C, 65.29; H, 7.53. Found: c, 65.50; H, 7.66. (-)-15: $[\alpha]_{20}^{20}$ -121 °(c 1.09, CHCl₃). IR (KBr): 1775, 1725 cm⁻¹. NMR δ : 1.18 (9H, s, (CH₃)₃), 1.24 (3H, s, CH₃), 2.0-2.6 (4H, m, (CH₂)₂), 2.6-3.1 (3H, m, CHCHCH), 4.06, 4.36 (1H, dd, J=3, 14 Hz, OCH₂CH), 4.65 (1H, t, J=3 Hz, CH). MS m/z 294 (M⁺). Anal. Calcd for C₁₆H₂₂O₅: C, 65.29; H, 7.53. Found: C, 65.14; H, 7.53.

(+)-(1R,2S,5R,6R,7S)-6-Methyl-5-pivaloyloxymethyl-4-oxatricyclo[5.3.0.0^{2,6}]decan-3,10dione 14 A mixture of above crude ketal (1.40 g) and 10% H₂SO₄ (1.7 mL) in acetone (27 mL) was refluxed for 1 h. Work up as described above afforded (+)-14 as colorless needles (1.22 g, quant).

(-)-(15,2R,5R,6S,7R)-6-Methyl-5-pivaloyloxymethyl-4-oxatricyclo[5.3.0.0^{2,6}]decan-3,10-dione Similarly, the compound was prepared from the above ketal as colorless needles (93%) of mp 161-163 °C (AcOEt-hexane). $[\alpha]_{20}^{20}$ -218 °(c 0.93, CHCl3). IR (KBr): 1780, 1730 cm⁻¹. NMR δ : 1.24 (9H, s, (CH3)3), 1.31 (3H, s, CH3), 1.8-2.2, 2.2-2.6, 2.7-2.9 (each 2H, m), 3.1-3.3 (1H, m), 4.34 (3H, s, OCH2CHO). MS m/z: 294 (M⁺). Anal. Calcd for C₁₆H₂₂O₅: C, 65.29; H, 7.53. Found: c, 65.02; H, 7.58.

(+)-(1R,2S,5R,6R,7S)-6-Methyl-10-methylene-5-pivaloyloxymethyl-4-oxatricyclo-

[5.3.0.0^{2,6}]decan-3-one 16 *n*-BuLi (1.50 M in hexane, 0.22 mL, 0.33 mmol) was added to a suspension of methyltriphenylphosphonium bromide (121 mg, 0.34 mmol) in DME (1.5 mL) at 0 °C, and the mixture was stirred at 0 °C for 1 h. Then the supernatant (1.3 mL, 0.25 mmol) was added to a solution of (+)-14 (50 mg, 0.17 mmol) in DME (4.5 mL) at -10 °C. The whole was stirred at rt for 1 h, and a mixture of satd NH4Cl and satd. NaCl was added. After extraction with CH₂Cl₂ (10 mL x 3), the combined extracts were washed with satd. NaCl, and dried. Concentration and column chromatography (ether-benzene 1:20) afforded (+)-16 (30 mg, 60%) as a colorless oil. $[\alpha]_{2D}^{2D} +79.4$ °(c 1.14, CHCl₃). IR (neat): 1775, 1735 cm⁻¹. NMR δ : 1.16 (3H, s, CH₃), 1.18 (9H, s, (CH₃)₃), 1.7-2.0 (2H, m), 2.4-2.7 (3H, m), 2.8-3.0, 3.1-3.3 (each 1H, m), 4.06, 4.34 (each 1H, dd, J=3, 12 Hz, OCH₂CH), 4.56 (1H, t, J=3 Hz, CHO), 5.0-5.1 (2H, m, C=CH₂). MS m/z: 292 (M⁺). HRMS m/z: Calcd for C₁₇H₂₄O₄ (M⁺): 292.1674. Found: 292.1676.

(+)-(1*R*,2*S*,5*R*,6*R*,7*S*,10*R*)-6,10-Dimethyl-5-pivaloyloxymethyl-4-oxatricyclo[5.3.0.0^{2,6}]decan-3-one 17 A solution of (+)-16 (1.40 g) and acetic acid (5 mL) in EtOH (50 mL) was hydrogenated over 10% Pd-C (0.2 g) under H₂ at rt for 22 h. After filtration, the filtrate was neutralized with satd. NaHCO₃, concentrated, and extracted with CH₂Cl₂ (50 mL x 3). The extract was dried. Concentration afforded (+)-17 (1.40 g, quant.) as a colorless oil. $[\alpha]_{2}^{24}$ +1.53 °(c 1.05, CHCl₃). IR (neat): 1775, 1730 cm⁻¹. NMR δ : 1.03 (3H, s, CH₃), 1.06 (3H, d, J=7 Hz, CH₃), 1.20 (9H, s, (CH₃)₃), 1.5-2.2 (5H, m), 2.45 (1H, d, J=5 Hz, CHCO), 2.62 (1H, t, J=7 Hz, CH₂CH), 2.82 (1H, dd, J=5, 7 Hz, CHCH), 4.12 (1H, dd, J=6, 12 Hz, OCH₂CH), 4.34 (1H, dd, J=4, 12 Hz, OCH₂CH), 4.70 (1H, dd, J=4, 6 Hz, CHOCO). MS m/z: 294 (M⁺). HRMS m/z: Calcd for C₁₇H₂₆O₄ (M⁺): 294.1831. Found: 294.1856.

(+)-(1R,2S,6S,7S,10R)-10,6-Dimethyltricyclo[5.3.0.0^{2,6}]dec-4-en-3-one 9 MeLi (1.40 M in ether, 4.5 mL, 6.3 mmol) was added to a solution of (+)-17 (1.42 g, 4.8 mmol) in THF (35 mL) at -78 °C, and the mixture was stirred at -78 °C for 20 min, and MeOH (10 mL) was added. The whole was heated under reflux for 1 h. After concentration, water (15 mL) and AcOEt (30 mL) were added, and the mixture was stirred at rt for 30 min. Then NaIO4 (3.0 g, 14 mmol) was added, and the mixture was stirred at rt for 2 h. After extraction with AcOEt (15 mL x 2), the combined extracts were washed with water, satd. Na2S₂O₃, and satd. NaCl, then dried. Concentration afforded the ketoaldehyde (1.0 g) as a colorless oil. A solution of ketoaldehyde (1.05 g) in MeOH (22 mL) and 15% NaOH (8.6 mL) was stirred at rt for 1 h, and water (50 mL) was added. After extraction with CH₂Cl₂ (20 mL x 3), the combined extracts were dried. Concentration and column chromatography (AcOEt-hexane 1:10) afforded (+)-9 (0.65 g, 76%) as a colorless oil. [α]²_D +214 °(c 1.78, CHCl₃). IR (neat): 1700, 1580 cm⁻¹. NMR δ : 1.06 (3H, s, CH₃), 1.07 (3H, d, J=7 Hz, CH₃), 1.2-2.6 (8H, m), 6.16, 7.51 (each 1H, d, J=7 Hz, CH=CH). MS m/z: 176 (M⁺). HRMS m/z: Calcd for C₁2H₁₆O (M⁺): 176.1199. Found: 176.1186.

(-)-(1R,2S,5R,6R,7S)-6-Methyl-5-pivaloyloxymethyl-4-oxatricyclo[5.3.0.0^{2,6}]dec-9-ene-3,8dione 18 A solution of (-)-15 (2.66 g, 9.05 mmol) and phenylselenyl chloride (2.80 g, 14.6 mmol) in AcOEt (50 mL) was stirred at rt for 1.5 h and concentrated. To a solution of the residue and pyridine (6 mL) in CH₂Cl₂ (100 mL) was added 15% H₂O₂ (3.6 mL, 16 mmol) at 0 °C, and the whole was stirred at 0 °C for 1.5 h, washed with water, satd. Na₂S₂O₃, water, 10% HCl, water, and satd. NaHCO₃, then dried. Concentration and chromatography (ether-benzene 1:3) afforded (-)-18 (1.77 g, 69%) as colorless needles of mp 125-126 °C (AcOEt-hexane). $[\alpha]_{24}^{24}$ -208 °C (c 0.42, CHCl₃). IR (KBr): 1770, 1730, 1690 cm⁻¹. NMR δ : 1.17 (9H, s, (CH₃)₃), 1.18 (3H, s, CH₃), 2.74 (1H, d, J=1 Hz, CH), 3.12 (1H, d, J=5 Hz, CHCOO), 3.5-3.6 (1H, m), 4.06 (1H, dd, J=2, 12 Hz, OCH₂CH), 4.39 (1H, dd, J=3, 12 Hz, OCH₂CH), 4.76 (1H, dd, J=2, 3 Hz, CHOCO), 6.49 (1H, dd, J=1, 5 Hz, CH=CHCO), 7.87 (1H, dd, J=3, 5 Hz, CH=CH). MS m/z 292 (M⁺). Anal. Calcd for C₁₆H₂₀O₅·1/4H₂O: C, 64.74; H, 6.96. Found: C, 64.83; H, 6.85.

(-)-(1R,2S,5R,6R,7S)-6,10-Dimethyl-5-pivaloyloxymethyl-4-oxatricyclo[5.3.0.0^{2,6}]dec-9ene-3,8-dione 19 N-Methyl-N-nitrosourea (12 g) was added to a mixture of ether (70 mL) and 40% KOH (35 mL) at 0 °C. Ether layer was added to a solution of (-)-18 (1.77 g, 6.06 mmol) in THF (35 mL). The whole was stirred at 0 °C for 3 h, and concentrated. A solution of the residue in toluene (100 mL) was heated under reflux for 1 h. Concentration and column chromatography (AcOEt-hexane 1:5) afforded (-)-19 (1.40 g, 75%) as colorless needles of mp 165-166 °C (AcOEt-hexane). $[\alpha]_D^{2}$ -162 °(c 0.87, CHCl3). IR (KBr): 1770, 1740, 1690 cm⁻¹. NMR δ : 1.18 (12H, s, (CH3)3 and CH3), 2.25 (3H, s, CH3C=C), 2.63 (1H, d, J=2 Hz, CH), 3.13 (1H, d, J=5 Hz, CHCOO), 3.31 (1H, dd, J=2, 5 Hz, CHCHCO), 4.05 and 4.40 (each 1H, dd, J=12, 3 Hz, OCH2CH), 4.75 (1H, t, J=3 Hz, CHOCO), 6.22 (1H, br, C=CHCO). MS m/z: 306 (M⁺). Anal. Calcd for C₁₇H₂₂O₅: C, 66.65; H, 7.24. Found: C, 66.41; H, 7.30.

(-)-(1R,2S,5R,6R,7S,10R)-6,10-Dimethyl-5-pivaloyloxymethyl-4-oxatricyclo[5.3.0.0^{2,6}]decan-3,8-dione A solution of (-)-19 (1.40 g) in AcOEt (30 mL) was hydrogenated over 10% Pd-C (100 mg) under H₂ at rt for 24 h. After filtration, the filtrate was concentrated to give colorless plates of mp 123-124 °C (AcOEt-hexane) (1.31 g, 93%). $[\alpha]_{25}^{25}$ -177 °(c 1.08, CHCl₃). IR (KBr): 1780, 1730 cm⁻¹. NMR δ : 1.10 (3H, s, CH₃), 1.21 (9H, s, (CH₃)₃), 1.22 (3H, d, J=6 Hz, CH₃), 2.1-2.9 (5H, m), 3.0-3.2 (1H, m), 4.17 (1H, dd, J=12, 6 Hz, OCH₂CH), 4.36 (1H, dd, J=12, 4 Hz, CH₂CH), 4.75 (1H, dd, J=4, 6 Hz, CHOCO). MS m/z: 309 (M⁺+1). Anal. Calcd for C₁₇H₂₄O₅: C, 66.21; H, 7.85. Found: C, 65.95; H, 7.88. (+)-(15,25,5*R*,65,75,10*R*)-8-Ethylenedithio-6,10-dimethyl-5-pivaloyloxymethyl-4-oxatricyclo[5.3.0.0^{2,6}]decan-3-one A solution of 19 (1.31 g, 4.3 mmol), ethanedithiol (2.5 mL, 30 mmol) and BF₃OEt₂ (0.2 mL) in CH₂Cl₂(50 mL) was stirred at rt for 2 days, then washed with water and satd. NaHCO₃, and dried. Concentration and chromatography (AcOEt-hexane 1:5) afforded colorless needles of mp 112.5-113 °C (AcOEt-hexane) (1.50 g, 92%). $[\alpha]_D^{24}$ +16.7 °(c 1.22, CHCl₃). IR (KBr): 1775, 1730 cm⁻¹. NMR δ : 1.11 (3H, d, J= 7 Hz, CH₃), 1.20 (9H, s, (CH₃)₃), 1.33 (3H, s, CH₃), 2.0-2.7 (4H, m), 2.8-3.0 (2H, m), 3.1-3.4 (4H, m, S(CH₂)₂S), 4.14, 4.35 (each 1H, dd, J=12, 4 Hz, OCH₂CH), 4.68 (1H, t, J=4 Hz, CH). MS m/z: 384 (M⁺). Anal. Calcd for C₁₉H₂₈O4S₂: C, 59.34; H, 7.34. Found: C, 59.19; H, 7.34.

(+)-(1*R*,2*S*,5*R*,6*R*,7*S*,10*R*)-6,10-Dimethyl-5-pivaloyloxymethyl-4-oxatricyclo[5.3.0.0^{2,6}]decan-3-one 17 A suspension of above dithioacetal (1.45 g), Raney Ni (W-4) (10 mL) in EtOH (100 mL) was heated under reflux for 2 h. Nickel was filtered off, and the filtrate was concentrated and purified by column chromatography (AcOEt-hexane 1:5) to afford (+)-17 (1.05 g, 95%) as a colorless oil. $[\alpha]_{24}^{24}$ +1.56 °(c 0.90, CHCl₃). Spectroscopic data were identical in all respects with those of (+)-17 prepared from 16.

2-Cyclohexyl-1-bromoprop-2-ene 21 Phosphorous tribromide (0.55 mL, 8.3 mmol) was added to a solution of 20 (1.40 g, 10 mmol) in ether (10 mL) at 0 °C and the mixture was stirred at 0 °C for 1.5 h. The mixture was washed with satd. NaHCO₃, water, and satd NaCl, then dried. Concentration afforded 21 as a colorless oil (0.95 g, 47%). IR (neat): 3100, 2950, 2850, 1640 cm⁻¹. NMR δ : 0.8-2.5 (11H, m), 3.95 (2H, s, CH₂Br), 4.90 and 5.07 (each 1H, s, C=CH₂). MS m/z: 202 (M⁺) 123 (M⁺-Br).

2-Cyclohexyl-1-methylthioprop-2-ene 22a A 15% solution of sodium thiomethoxide in water (5 mL, 11 mmol) was added to a solution of 21 (0.94 g, 4.6 mmol) in ethanol (10 mL). The mixture was stirred at rt of 4.5 h, and taken up into ether, washed with satd. NaCl, and dried. Concentration and distillation afforded 22a as a colorless oil (0.79 g, 85%) of bp 150 °C/12 mmHg. IR (neat): 3100, 1640 cm⁻¹. NMR δ : 1.0-2.8 (11H, m), 1.95 (3H, s, SCH₃), 3.10 (2H, s, CH₂S), 4.80 (2H, s, C=CH₂). MS m/z: 170 (M+).

2-Cyclohexyl-1-ethylthioprop-2-ene 22b Ethanethiol (0.88 mL, 12 mmol) was added to a solution of sodium ethoxide in EtOH (10 mL, 12.5 mmol) followed by a solution of 21 (2.4 g, 12 mmol) in EtOH (5 mL). The mixture was stirred at rt for 1 h, taken up into ether, and washed with satd. NaCl, then dried. Concentration and distillation afforded 22b as a colorless liquid (1.22 g, 56%) of bp. 130 °C/5 mmHg. IR (neat): 3100, 2950, 2850, 1635 cm⁻¹. NMR &: 0.8-2.5 (11H, m), 1.22 (3H, t, J=7 Hz, CH₃), 2.45 (2H, q, J=7 Hz, CH₂), 3.18 (2H, s, CH₂S), 4.84 (2H, s, C=CH₂). MS m/z: 184 (M⁺).

Preparation of sulfonium salt 23b, generation of ylide, and coupling with 24 (Table, entry 5) Et30.BF4 (99 mg, 0.52 mmol) was added to a solution of 22b (85 mg, 0.46 mmol) in CH2Cl2 (2 mL) and the mixture was stirred at rt for 3 h. After concentration, the residual sulfonium salt (23b) was dissolved in DME (3 mL) and cooled to -78 °C. After addition of n-BuLi (1.50 M in hexane, 0.36 mL, 0.50 mmol), the mixture was stirred for 10 min. A solution of (±)-24 (20% in ether, 0.3 mL, 0.6 mmol) in DME (1 mL) was added at -78 °C, and the mixture was stirred at -78 °C for 1 h and at rt for 1 h. After addition of water, the mixture was extracted with ether (30 mL). The combined extracts were dried. Concentration and chromatography (AcOEt-hexane 1:50) gave a 1:1 mixture of 25 and one isomer of 27 as a colorless oil (22 mg, 22%), 26 as a colorless oil (5 mg, 5%), another isomer of 27 as a colorless oil (6 mg, 6%), and 28b as a colorless oil (19 mg, 19%). 25 and one isomer of 27: IR (neat): 3100, 2950, 1640 cm⁻¹. NMR δ: 0.9-2.1 (17H, m), 2.52 (0.5H, d, J=8 Hz, CH(O)CMe2 of 25), 2.61 (0.5H, d, J=6 Hz, CH(O)CMe2 of 27), 2.67 (0.5H, dd, J=2, 6 Hz, C=CCH(O)CH of 27), 2.90 (0.5H, d, J=5, 8 Hz, C=CCH(O)CH of 25), 3.36 (0.5H, dd, J=2, 1 Hz, C=CCH(O)CH of 27), 3.54 (0.5 H, d, J=5, 1 Hz, C=CCH(O)CH of 25), 4.88, 4.98 (each 0.5 H, s, C=CH2), 5.01 (1H, s, C=CH2). MS m/z: 233 (M++1), 207 (M+-CH3). 26: IR (neat): 3100, 2950. 1650 cm⁻¹. NMR 8: 1.0-1.5 (5H, m), 1.28, 1.36 (each 3H, s, CH3), 1.6-2.1 (6H, m), 2.53 (1H, d, J=8 Hz, CH(O)CMe2), 2.85 (1H, dd, J=8, 5 Hz, CH(O)CH), 3.48 (1H, dd, J=5, 1 Hz, CH(O)CH), 4.96, 5.07 (each 1H, s, C=C<u>H</u>₂). MS m/z: 222 (M⁺). Another isomer of 27: IR (neat): 2950 cm⁻¹. NMR δ : 1.0-1.5 (5H, m), 1.36, 1.41 (each 3H, s, CH3)), 1.6-2.0 (6H, m), 2.63 (1H, d, J=6 Hz CH(O)CMe2), 2.72 (1H, dd, J=2, 6 Hz, CH(O)CH), 3.29 (1H, d, J=2 Hz, CH(O)CH), 4.89, 5.04 (each 1H, s, C=CH2). MS m/z: 222 (M+). 28b: IR (neat): 3100, 1640 cm⁻¹. NMR δ: 1.0-1.4 (5H, m), 1.4-2.0 (6H, m), 1.24 (3H, t, J=7 Hz, CH₃), 1.22 (3H, d, J=6 Hz, CH3), 2.10 (1H, dd, J=5, 9 Hz, C=CCH2), 2.44 (1H, dd, J=5, 13 Hz, C=CCH2), 2.7-3.0 (1H, m, SCH), 2.58 (2H, q, J=7 Hz, SCH2), 4.6-4.9 (2H, m, C=CH2). MS m/z: 212 (M+).

Preparation of sulfonium salt 23a, generation of ylide, and coupling with 24 (Table, entry 2) Me₃O·BF₄ (62 mg, 0.42 mmol) was added to a solution of 22a (65 mg, 0.38 mmol) in CH₂Cl₂ (2 mL) and the mixture was stirred at rt for 40 min. After concentration, the residual sulfonium salt (23a) was treated as described above affording a 1:1 mixture (25 and one isomer of 27) (7 mg, 8%), another isomer of 27 (0.7 mg, 1%), and 28a as a colorless oil (40 mg, 40%). 28a: IR (neat): 3100, 1640 cm⁻¹. NMR δ : 0.8-2.0 (11H, m), 2.10 (3H, s, SCH₃), 2.2-2.9 (4H, m), 4.6-4.8 (2H, m, C=CH₂). MS m/z: 184 (M⁺).

Preparation of sulfonium salt 23c, generation of ylide, and coupling with 24 (Table, entry 4) A solution of 20 (1.15 g, 8.2 mmol), methanesulfonyl chloride (0.76 mL, 9.8 mmol), and triethylamine (1.6 mL, 10.8 mmol) in CH₂Cl₂ (20 mL) was stirred at 0 °C for 1.5 h, and then washed with 10% HCl, water, and satd. NaHCO₃, then dried. Concentration afforded mesylate of 20 (1.83 g, quant.) as a colorless oil. A solution of mesylate (0.50 g) and sodium iodide (0.66 g, 4.4 mmol) in acetone (5 mL) was stirred at rt for 1 h, taken up into ether, and the mixture was washed with 10% Na₂S₂O₃, water, and satd. NaCl, then dried. Concentration gave the corresponding iodide as a pale yellow oil (0.44 g, 77%). To a solution of iodide (99 mg, 0.40 mmol) and diphenyl sulfide (0.65 mL, 3.9 mmol) in nitromethane (1 mL) was added AgBF₄ (92 mg, 0.47 mmol), and the mixture was stirred at rt for 2.5 h. After addition of CH₂Cl₂ (5 mL), the mixture was filtered, concentrated, taken up into ether, and the supernatant was decanted off. The residue was dried under reduced pressure affording 23c as a colorless oil (88 mg). Generation of ylide and following coupling afforded 26 (1 mg, 2%), and a 1:1 mixture of 25 and 27 (4 mg, 7%).

(3S,4S,5S)-2-Cyclohexyl-5,6-epoxy-4-hydroxy-6-methyl-3-methylthiohept-1,5-diene 30 *n*-BuLi (1.50 M in hexane, 0.74 mL, 1.1 mmol) was added to a solution of 22a (170 mg, 1.0 mmol) and HMPA (0.35 mL, 2 mmol) in THF (8 mL) at -20 °C, and the mixture was stirred at -20 °C for 30 min. Tri*n*-butyltin chloride (0.30 mL, 1.1 mmol) was added at -70 °C. The mixture was stirred at rt for 15 min and quenched with water. The mixture was extracted with ether. The extract was washed with water and brine, then dried. Concentration afforded 29 as a colorless oil (0.54 g). BF3·OEt2 (0.05 mL, 0.42 mmol) was added to a solution of 29 (207 mg) and 24 (20% in ether, 0.36 mL, 0.72 mmol) in CH₂Cl₂ (4 mL) at -78 °C, stirred at -78 °C for 5 min, and quenched by satd. NaHCO3. The mixture was extracted with ether. The extract was washed with satd. NaCl, and dried. Concentration and chromatography (ether-benzene 1:1) afforded 30 as a colorless oil (21 mg, 20% from 22a). IR (neat): 3440, 1635 cm⁻¹. NMR & 0.8-2.0 (17H, m), 2.05 (3H, s, SCH₃), 2.62 (1H, d, J=3 Hz, OH), 2.81 (1H, d, J=8 Hz, SCH), 3.20 (1H, d, J=8 Hz, CH(O)CMe2), 3.6) (1H, dt, J=3, 8 Hz, CHOH), 4.90, 5.05 (each 1H, s, C=CH₂). MS m/z: 270 (M⁺).

Transformation of 30 to 26 A solution of 30 (6 mg, 0.022 mmol) and Me₃O·BF₄ (4 mg, 0.027 mmol) in CH₂Cl₂ (0.5 mL) was stirred at rt for 2 h. After concentration, the residue was dissolved in DME (1 mL). HMPA (3.8 μ L, 0.022 mmol) and MeLi (0.86 M in ether, 0.03 mL, 0.026 mmol) were added at -78 °C, and the mixture was stirred at -78 °C for 40 min and then at rt for 1.5 h, and quenched by pH 7 phosphate buffer, then extracted with CH₂Cl₂. The extract was dried. Concentration and column chromatography afforded 26 as a colorless oil (0.5 mg, 10%).

(-)-(1S,2R,3R,5R,6R,7S,8R)-5-(1-Ethylthio-2-propen-2-yl)-3-methoxymethoxy-2,8-di-

methyltricyclo[5.3.0.0^{2,6}]decane A solution of (-)-10 (311 mg, 1.11 mmol), methanesulfonyl chloride (0.16 mL, 2.1 mmol) and triethylamine (0.40 mL, 2.9 mmol) in CH₂Cl₂ (10 mL) was stirred at 0 °C for 30 min and diluted with CH₂Cl₂ (20 mL), then washed with 5% HCl, water, and satd. NaHCO₃, then dried. Concentration afforded mesylate of (-)-10 (413 mg, quant.) as a colorless oil. Sodium thioethoxide (0.67 M in DMF, 1.7 mL, 1.2 mmol) was added to a solution of above mesylate (413 mg, 1.11 mmol) in DMF (3 mL) at 0 °C, and the whole was stirred at rt for 20 min, and diluted with benzene (50 mL), then washed with water and satd. NaCl, then dried. Concentration and chromatography (AcOEt-benzene 1:100) gave a colorless oil (319 mg, 89%). $[\alpha]_{20}^{20}$ -2.56°(c 0.86, CHCl₃). IR (neat): 3100, 1630 cm⁻¹. NMR δ : 0.88 (3H, d, J=7 Hz, CH₃), 1.00 (3H, s, CH₃), 1.1-1.5 (2H, m), 1.23 (3H, t, J=7 Hz, CH₃), 1.5-2.2 (7H, m), 2.2-2.7 (3H, m), 2.9-3.4 (3H, m), 3.36 (3H, s, OCH₃), 3.62 (1H, d, J=4 Hz, CHOCH₂), 4.56, 4.68 (each 1H, d, J=7 Hz, OCH₂O), 4.87 (1H, t, J=2 Hz, CH₂=C), 4.98 (1H, br, CH₂=C). MS m/z: 324 (M⁺).

(-)-(1*S*,2*R*,3*R*,5*R*,6*R*,7*S*,8*R*)-5-(1-Ethylthio-2-propen-2-yl)-3-hydroxy-2,8-dimethyltricyclo-[5.3.0.0^{2,6}]decane 31 A solution of above sulfide (88 mg, 0.27 mmol) and 10% H₂SO₄ (0.5 mL) in acetone (5 mL) was heated under reflux for 2.5 h. After dilution with benzene (50 mL), the mixture was washed with satd. NaHCO₃ and satd. NaCl, then dried. Concentration and chromatography (AcOEt-benzene 1:10) gave (-)-31 (70 mg, 92%) as colorless needles of mp 101-102.5 °C (hexane). $[\alpha]_D^{24}$ +49.6°(c 0.73, CHCl₃). IR (KBr): 3300, 1630 cm⁻¹. NMR & 0.88 (3H, d, J=8 Hz, CH₃), 1.00 (3H, s, CH₃), 1.0-2.1 (10H, m), 1.22 (3H, t, J=7 Hz, CH₃), 2.27 (1H, dt, J=4, 13 Hz, CH₂), 2.44 (2H, q, J=7 Hz, SCH₂), 2.96 and 3.16 (each 1H, d, J=14 Hz, SCH₂C=CH₂), 3.1-3.5 (1H, m), 3.75 (1H, d, J=4 Hz, HOCH), 4.87 (1H, t, J=2 Hz, C=CH₂), 4.98 (1H, t, J=1 Hz, C=CH₂). MS m/z: 280 (M⁺): 280 (M⁺). Anal. Calcd for C₁₇H₂₈OS: C, 72.80; H, 10.06. Found: C, 72.66; H, 10.20.

(+)-Diethyl 2-((1*S*,2*R*,3*R*,5*R*,6*R*,7*S*,10*R*)-5-hydroxy-6,10-dimethyltricyclo[5.3.0.0^{2,6}]decan-3-yl)prop-2-en-1-yl sulfonium tetrafluoroborate 11 A solution of (-)-31 (9 mg, 0.032 mmol) and Et₃O·BF₄ (6.7 mg, 0.035 mmol) in CH₂Cl₂ (0.5 mL) was stirred at rt for 70 h. After concentration, the residue was crystallized from ether to give (+)-11 (9.6 mg, 75%) as colorless plates of mp 123-126 °C (CH₂Cl₂-ether). Concentration of the mother liquor afforded the recovered (-)-31 (2 mg, 22% recovery). 11: $[\alpha]_D^{2}$ +84.2 °(c 0.24, CHCl₃). IR (KBr): 3300, 1625 cm⁻¹. NMR & 0.89 (3H, d, J=5 Hz, CH₃), 1.02 (3H, s, CH₃), 1.1-2.1 (11H, m), 1.50, 1.53 (each 3H, t, J=7 Hz, S(CH₂CH₃)₂), 2.28 (1H, dt, J=4, 13 Hz, CH₂CH), 2.9-3.2 (1H, m, CHC=CH₂), 3.37 and 3.40 (each 1H, q, J=7 Hz, S(CH₂CH₃)₂), 3.48 (2H, q, J=7 Hz, S(CH₂CH₃)₂), 3.78 and 4.15 (each 1H, d, J=13 Hz, CH₂S), 3.80 (1H, d, J=5 Hz, HOCH), 5.35 (1H, d, J=2 Hz, CH₂=C), 5.58 (1H, s, CH₂=C). Anal. Calcd for C₁9H₃₃OSBF₄·1/4H₂O: C, 56.71; H, 8.43. Found: C, 56.64; H, 8.39.

(2R)-2,3-Epoxy-3-methylbutanal 24 CrO₃ (14.4 g, 0.13 mol) was added to a solution of pyridine (17 mL, 0.22 mol) in CH₂Cl₂ (200 mL) at 0 °C and the mixture was stirred at rt for 30 min. After addition of dry celite (70 g), (2S)-2,3-epoxy-3-methylbutan-1-ol¹² (2.4 g, 14 mmol) was added. The mixture was stirred at rt for 20 min. After addition of a mixture of ether and pentane (1:1, 170 mL), MgSO₄ (40 g), and NaHSO₄·H₂O (40 g), the mixture was filtered through florisil. The filtrate was concentrated to about 20 mL. Distillation (rt/3 mmHg) gave 40% ether solution of 24 (2.5 g, 42%). NMR δ : 1.38, 1.42 (each 3H, s, CH₃), 3.10 (1H, d, J=6 Hz, CHCHO), 9.30 (1H, d, J=6 Hz, CHCHO).

Generation of ylide from (+)-11 and coupling with 24 providing (+)-1 MeLi (0.86 M in ether, 0.80 mL, 0.69 mmol) was added to a solution of (+)-11 (111 mg, 0.28 mmol) and HMPA (0.053 mL, 0.30 mmol) in DME (20 mL) at -70 °C, and the mixture was stirred at -70 °C for 1 h. To this solution, 24 (40% in ether, 0.3 g, 1.2 mmol) in DME (1 mL) was added at -70 °C, the whole was stirred at -70 °C for 50 min, at rt for 1 h, and cooled to -50 °C. After addition of pH 7 phosphate buffer, the mixture was extracted with CH₂Cl₂ (20 mL x 3), and the combined extracts were dried. Concentration and chromatography (AcOEthexane 1:3) gave 34 (20 mg, 35%) as a colorless oil and the mixture of 1, 32, and 33 (18 mg) as an oil. Separation of the mixture of 1, 32, and 33 by HPLC (Waters Radial Pak B, acetone-hexane 1:6) afforded (+)-1 (2 mg, 3%) as colorless solid, (+)-32 (1.5 mg, 1.5%) as a colorless oil, and (+)-33 (5 mg, 8%) as a colorless oil. (+)-1: mp 100-101 °C (colorless needles from petroleum ether). $[\alpha]_{D}^{23}$ +45.5 °(c 1.23, CHCl3). IR (nujol): 3360, 1635 cm⁻¹. NMR δ : 0.93 (3H, d, J=6 Hz, CH₃), 1.01, 1.32, 1.43 (each 3H, s, CH₃), 1.1-2.2 (10H, m), 2.31 (1H, dt, J=4, 13 Hz, CH2CHC=CH2), 2.51 (1H, d, J=8 Hz, (CH3)2C(O)CH), 2.90 (1H, dd, J=8, 4 Hz, (CH3)₂C(O)CHCH(O)CH), 2.9-3.2 (1H, m), 3.47 (1H, dd, J=4, 0.8 Hz, (CH3)₂C(O)CHCH(O)CH), 3.76 (1H, d, J=4 Hz, HOCH), 5.05, 5.16 (each 1H, t, J=1.4 Hz, CH₂=C). MS m/z; 318 (M⁺). (+)-32: $[\alpha]_{13}^{23}$ +11 °(c 0.65, CHCl₃). IR (KBr): 3360 cm⁻¹. NMR &: 0.93 (3H, d, J=8 Hz, CH₃), 1.01, 1.29, 1.39 (each 3H, s, CH3), 1.1-2.4 (11H, m), 2.57 (1H, d, J=7 Hz, (CH3)₂C(0)CH), 2.94 (1H, dd, J=7, 4 Hz, (CH₃)₂C(O)CHC<u>H</u>(O)CH), 2.9-3.3 (1H, m), 3.51 (1H, dd, J=4, 1 Hz, (CH₃)₂C(O)CHCH(O)C<u>H</u>), 3.77 (1H, d, J=4 Hz, HOCH), 5.05 and 5.14 (each 1H, dd, J=1, 3 Hz, CH₂=C). MS m/z; 318 (M⁺). (+)-33: $[\alpha]_{23}^{23} + 24.0$ °(c 1.00, CHCl₃). IR (neat): 3440, 3100, 1640 cm⁻¹. NMR δ: 0.89 (3H, d, J=6 Hz, CH₃), 0.99,1.35, 1.40 (each 3H, s, CH3), 1.1-2.2 (11H, m), 2.57 (1H, d, J=7 Hz, (CH3)₂C(O)CH), 2.83 (1H, dd, J=7, 2 Hz, (CH₃)₂C(O)CHC<u>H</u>(O)CH), 2.8-3.1 (1H, m), 3.33 (1H, d, J=2 Hz, (CH₃)₂C(O)CHCH(O)C<u>H</u>), 3.75 (1H, d, J=4 Hz, HOC<u>H</u>), 4.97 and 5.20 (each 1H, s, C<u>H2</u>=C). MS m/z: 318 (M⁺). (+)-34: $[\alpha]_{1}^{24}$ +30.0 °(c 0.26, CHCl₃). IR (neat): 3440, 3080, 1635 cm⁻¹. NMR δ: 0.91 (3H, d, J=6 Hz, CH₃), 1.01 (3H, s, CH₃), 1.1-2.7 (21H, m), 2.8-3.2 (2H, m), 3.75 (1H, d, J=4 Hz, HOCH), 4.7-5.0 (2H, m). MS m/z; 308 (M⁺),

References and Notes

^{1.} Stereoselective Reactions. Part 26. For the part 25, see: Tanaka, M.; Tomioka, K.; Koga, K. preceding paper.

- 2. Gerwick, W. H.; Fenical, W.; Engen, D. V.; Clardy, J. J. Am. Chem. Soc. 1980, 102, 7991.
- Fernandes, S. L.; Kamat, S. Y. Tetrahedron Lett. 1980, 21, 2249; Gerwick, W. H.; Fenical, W.; Sultanbawa, M. U. S. J. Org. Chem. 1981, 46, 2233; Silva, S. S. M. D.; Gamage, S. K. T.; Kumar, N. S.; Balasubramanian, S. Phytochemistry 1982, 21, 944; Ravi, B. N.; Wells, R. J. Aust. J. Chem. 1982, 35, 129; Gerwick, W. H.; Fenical, W. J. Org. Chem. 1983, 48, 3325.
- Krepinsky, J.; Samek, Z.; Sorm, F.; Lamparsky, D.; Ochsner, P.; Navas, Y.-R. Tetrahedron 1967, S8, 53; Gianotti, C.; Schwang, H. Bull. Soc. Chim. France 1968, 2452.
- Linear vinylic diepoxide (hedamycin): Ceroni, M.; Seuin, U. Tetrahedron Lett. 1979, 3703. Cyclic diepoxide (coriamyrtin): Tanaka, K.; Uchiyama, F.; Sakamoto, K.; Inubushi, Y. J. Am. Chem. Soc. 1982, 104, 4965; Niwa, H.; Wakamatsu, K.; Hida, T.; Niiyama, K.; Kigoshi, H.; Yamada, M Nagase, H.; Suzuki, M.; Yamada, K. ibid. 1984, 106, 4547.
- Synthesis of spatanes: Salomon, R. G.; Sachinvala, N. D.; Raychaudhari, S. R.; Miller, D. B. J. Am. Chem. Soc. 1984, 106, 2211; Salomon, R. G.; Basu, B.; Roy, S.; Sharma, R. B. Tetrahedron, Lett. 1989, 30, 4621; Dauben, W. G.; Kowalczyk, B. A. *ibid.* 1990, 31, 635; Salomon, R. G.; Basu, B.; Roy, S.; Sachinvala, N. D. J. Am. Chem. Soc. 1991, 113, 3096. Synthesis of bourbonenes: White, J. D.; Gupta, D. N. *ibid.* 1968, 90, 6171; Brown, M. J. Org. Chem. 1968, 33, 162; Heathcock, C. H.; Bodger, R. A. J. Chem. Soc. Chem. Commun. 1968, 1510; Yoshihara, K.; Ohta, Y.; Sakai, T.; Hirose, Y. Tetrahedron Lett. 1969, 2263; Uyehara, T.; Ohnuma, T.; Sato, T.; Kato, K. J. Chem. Soc. Chem. Commun. 1981, 127.
- Bourbonene: Tomioka, K.; Tanaka, M.; Koga, K. Tetrahedron Lett. 1982, 23, 3401; Tomioka, K.; Tanaka, M.; Koga, K. Chem. Pharm. Bull. 1989, 37, 1201. Stoechospermol: Tanaka, M.; Tomioka, K.; Koga, K. Tetrahedron Lett. 1985, 26, 3035.
- Review: Baldwin, S. M. Organic Photochemistry 1981, 5, 123. Asymmetric (2+2) photocycloadditions: Partridge, J. J; Chadha, N. K.; Uskokovic, M. R. J. Am. Chem. Soc. 1973, 95, 532; Williams, J. R.; Callahan, J. F. J. Org. Chem. 1980, 45, 4475, 4479; Baldwin, S. W.; Crimmins, M. T. J. Am. Chem. Soc. 1982, 104, 1132; Tolbert, L. M.; Ali, M. B. ibid. 1982, 104, 1742; Jarosz, S.; Zamojski, A. Tetrahedron 1982, 38, 1447, 1453; Bruneel, K.; Keukeleire, D. D.; Vandevelle, M. J. Chem. Soc. Perkin Trans. 1 1984, 1967; Salomon, R. G.; Sachinvala, J. D.; Raychaundhuri, S. R.; Miller, D. B. J. Am. Chem. Soc. 1984, 106, 2211; Lange, G. L.; Decicco, C.; Tan, S. L.; Chamberlain, G. Tetrahedron, Lett. 1985, 26, 4707; Lange, G. L.; Lee, M. ibid. 1985, 26, 6163; Herzol, H.; Koch, H.; Sharf, H.-D.; Runsink, J. Tetrahedron 1986, 42, 3547; Meyers, A. I.; Fleming, S. A. J. Am. Chem. Soc. 1986, 106, 306; Lange, G. L.; Decicco, C. P. Tetrahedron, Lett. 1988, 29, 2613.
- Tomioka, K. Yakugakuzasshi 1984, 104, 1009; Tomioka, K.; Kawasaki, H.; Iitaka, Y.; Koga, K. Tetrahedron Lett. 1985, 26, 903; Tomioka, K.; Sugimori, M.; Koga, K. Chem. Pharm. Bull. 1986, 34, 1501; Tomioka, K.; Cho, Y.-S.; Sato, F.; Koga, K. J. Org. Chem. 1989, 53, 4094.
- 10. Tanaka, M.; Tomioka, K.; Koga, K. Tetrahedron Lett. 1985, 26, 6109.
- 11. Corey, E. J.; Chaykovsky, M. J. Am. Chem. Soc. 1965, 87, 1353; Trost B. M.; Martin, Jr., L. S. Sulfur Ylides, Academic Press, New York, 1975.
- 12. Yamada, S.; Shiraishi, M.; Ohmori, M.; Takayama, H. Tetrahedron Lett. 1984, 25, 3347.
- 13. Tanaka, M.; Tomioka, K.; Koga, K. Heterocycles, 1985, 23, 2347.
- Tomioka, K.; Sato, F.; Koga, K. Heterocycles 1982, 17, 311; Tomioka, K.; Ishiguro, T.; Iitaka, Y.; Koga, K. Tetrahedron 1984, 40, 1303.
- 15. Sharpless, K. B.; Lauer, R. F.; Teranishi, A. Y. J. Am. Chem. Soc. 1973, 95, 6137.
- 16. Marshall, A.; Anderson, N. H.; Hochstetler, A. R. J. Org. Chem. 1967, 32, 113.
- 17. Mihelich, E. D. Tetrahedron Lett. 1979, 4729.
- Yamamoto, Y.; Saito, Y.; Maruyama, K. J. Chem. Soc. Chem. Commun. 1982, 1326; Keck, G. E.; Abbott, D. E.; Boden, E. P.; Enholm, E. Tetrahedron Lett. 1984, 25, 3927.
- 19. Shimagaki, M.; Matsuzaki, Y.; Hori, I.; Nakata, T.; Oishi, T. Tetrahedron Lett. 1984, 25, 4779.
- 20. The trans-isomer was produced as a single isomer (the relative configuration was not determined)
- 21. The extract was dried over MgSO4. Silica gel chromatography was used. Melting points were measured using Büchi 510 melting point apparatus and are not corrected. Optical rotations were taken with a JASCO DIP-181 automatic polarimeter. Infrared spectra were taken with a JASCO Infrared spectrometer Model DS-402G and a JASCO IRA-I Grating Infrared Spectrometer. Proton nuclear magnetic resonance spectra were taken with a JEOL FX-100 Spectrometer at 100 MHz. CDCl₃ was used as a solvent. Chemical shifts are expressed in ppm relative to internal Me4Si. Abbreviations are as follows: s, singlet; t, triplet; q, quartet; m, multiplet; br, broad. Mass spectra (MS) were taken with a JEOL JMS DX-300 MS spectrometer.

(Received in Japan 31 August 1994; accepted 22 September 1994)